

# Batteries

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IB PHYSICS | ELECTRICITY

# Batteries



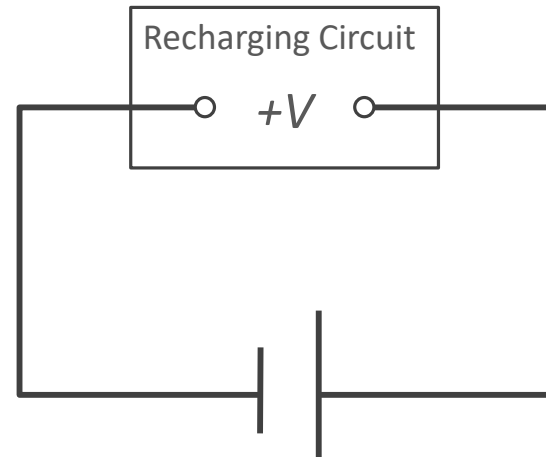
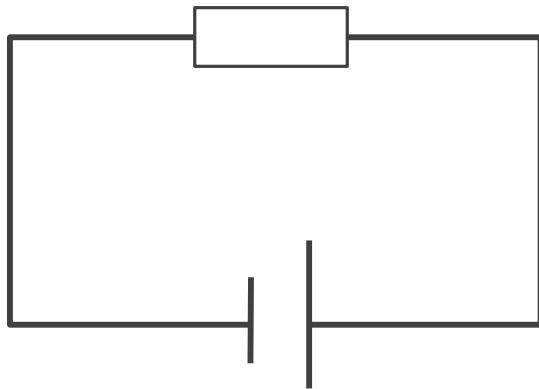
Primary Cells

Secondary Cells

Battery Shape	Chemistry	Nominal Voltage	Rechargeable?
AA, AAA, C, and D	Alkaline or Zinc-carbon	1.5V	No
9V	Alkaline or Zinc-carbon	9V	No
Coin cell	Lithium	3V	No
Silver Flat Pack	Lithium Polymer (LiPo)	3.7V	Yes
AA, AAA, C, D (Rechargeable)	NiMH or NiCd	1.2V	Yes
Car battery	Six-cell lead-acid	12.6V	Yes

# Recharging?

Some batteries can reverse the chemical reaction that produces the potential difference by passing a current through the battery in the opposite direction as it would normally travel

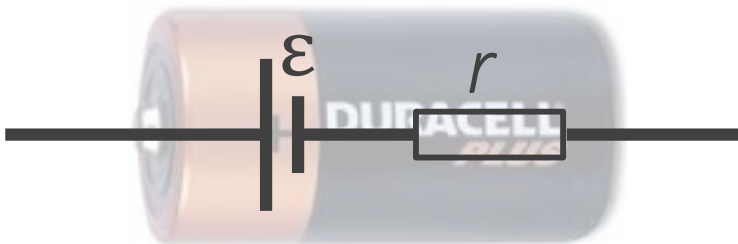


# Batteries | emf

We've been describing batteries so far as the voltage that they provide to the circuit, but that's not the whole story...

## Electromotive Force (emf)

The total energy transferred in the source per unit charge passing through it

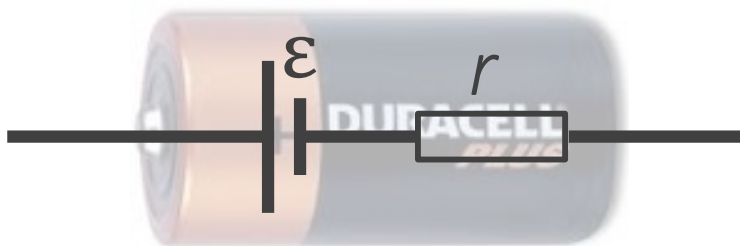


## Symbol

## Unit

# Batteries | Internal Resistance

All batteries have some amount of internal resistance

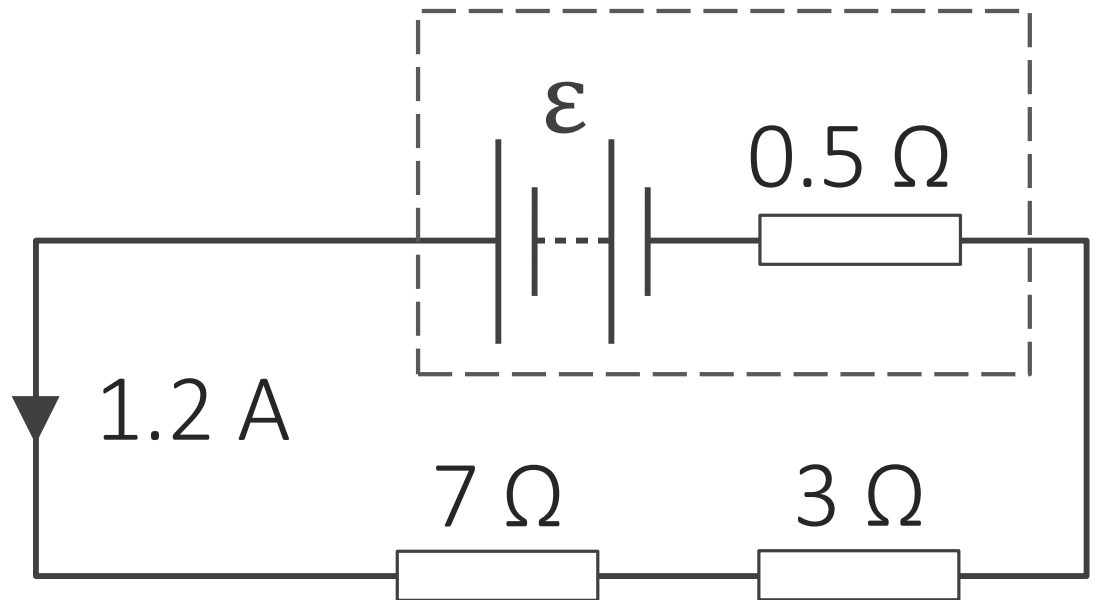


Symbol

Unit

# Batteries | emf

What is the emf for a battery shown below?



# IB Physics Data Booklet

## Sub-topic 5.1 – Electric fields

$$I = \frac{\Delta q}{\Delta t}$$

$$F = k \frac{q_1 q_2}{r^2}$$

$$k = \frac{1}{4\pi\epsilon_0}$$

$$V = \frac{W}{q}$$

$$E = \frac{F}{q}$$

$$I = nAvq$$

## Sub-topic 5.2 – Heating effect of electric currents

Kirchhoff's circuit laws:

$$\Sigma V = 0 \text{ (loop)}$$

$$\Sigma I = 0 \text{ (junction)}$$

$$R = \frac{V}{I}$$

$$P = VI = I^2 R = \frac{V^2}{R}$$

$$R_{\text{total}} = R_1 + R_2 + \dots$$

$$\frac{1}{R_{\text{total}}} = \frac{1}{R_1} + \frac{1}{R_2} + \dots$$

$$\rho = \frac{RA}{L}$$

## Sub-topic 5.3 – Electric cells

$$\mathcal{E} = I(R + r)$$

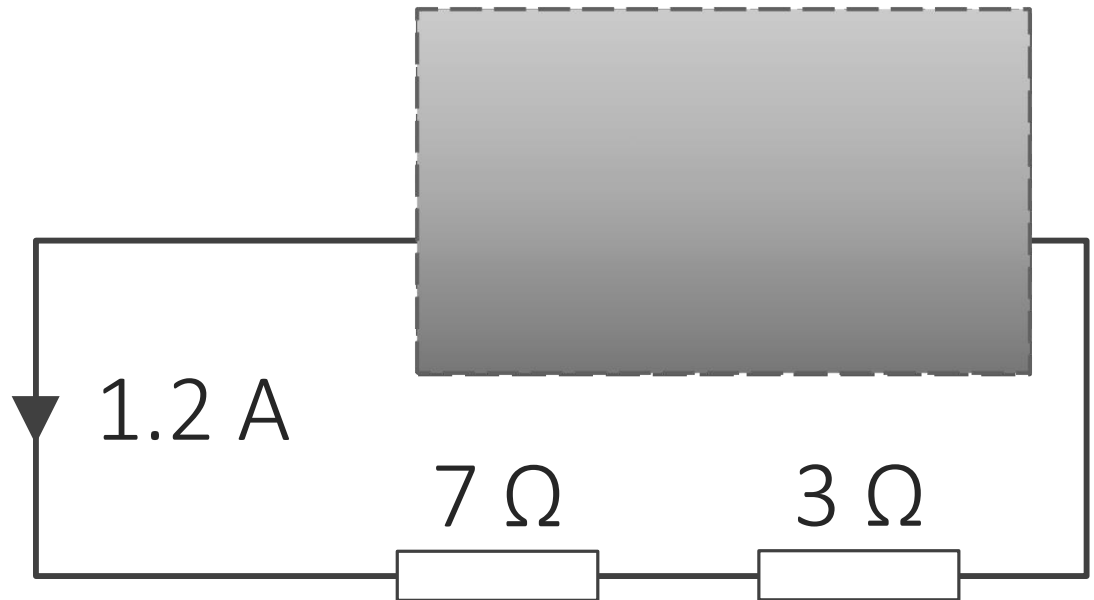
## Sub-topic 5.4 – Magnetic effects of electric currents

$$F = qvB \sin \theta$$

$$F = BIL \sin \theta$$

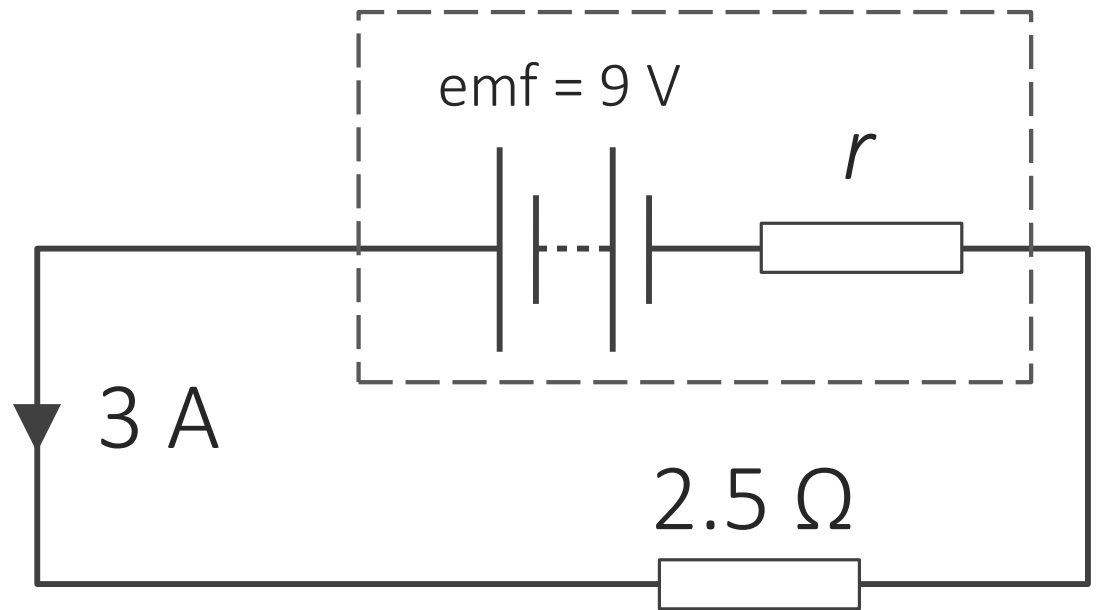
# Batteries | Terminal Voltage

What is the terminal voltage for a battery shown below?



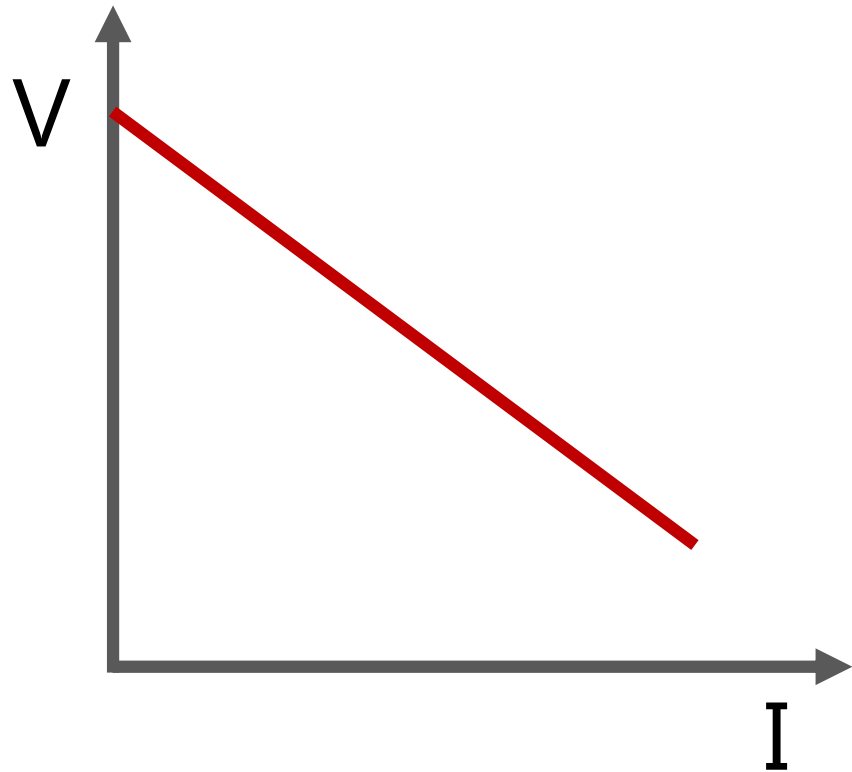
# Batteries | Internal Resistance

What is the internal resistance of this battery as shown below?



# Graphing Internal Resistance

$$\varepsilon = I(R + r)$$



# Lesson Takeaways

- ☐ I can describe the difference between primary and secondary cells
- ☐ I can define the electromotive force and describe how it is different than the battery's terminal voltage
- ☐ I can solve for a circuit that includes a battery with internal resistance
- ☐ I can describe how to determine emf and internal resistance from a Voltage vs Current graph